

Blood Gases Workshop 2021

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(Credits LITFL, WICM, Creative Commons)

Background

- Shortness of Breath is a common ED presentation and admission “*main complaint*”
- Respiratory Failure
 - leading cause of ICU admission
- Septic Shock
 - leading cause of mortality



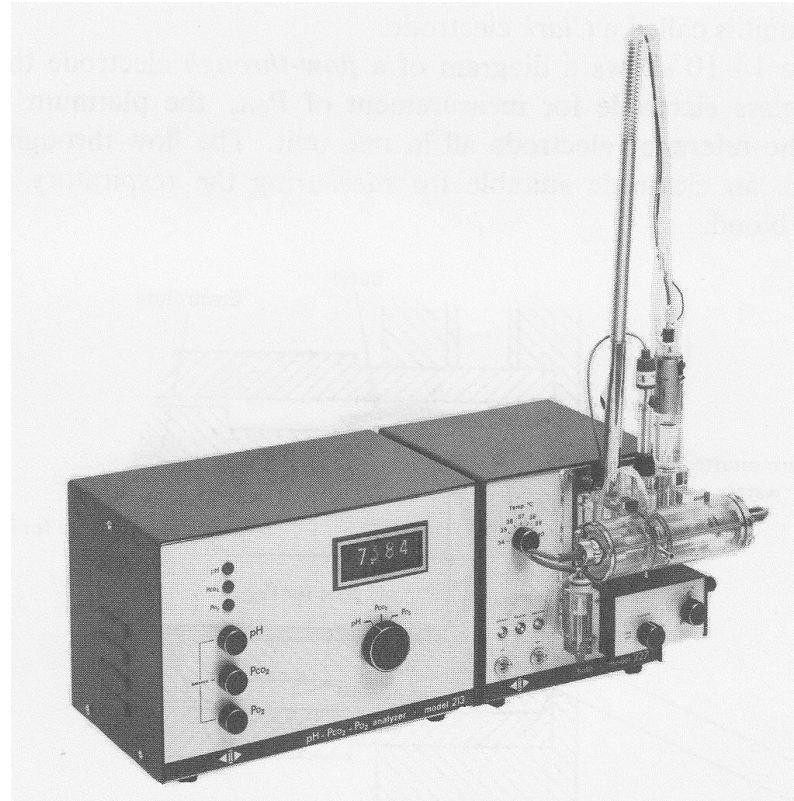
Agenda - Blood Gases Workshop

- Rationale, History, Types of Blood Gas
- Basic Rules Recap
- Advanced Rules
- Case Based Learning



Overview

- Historical Context



Why take a blood gas?

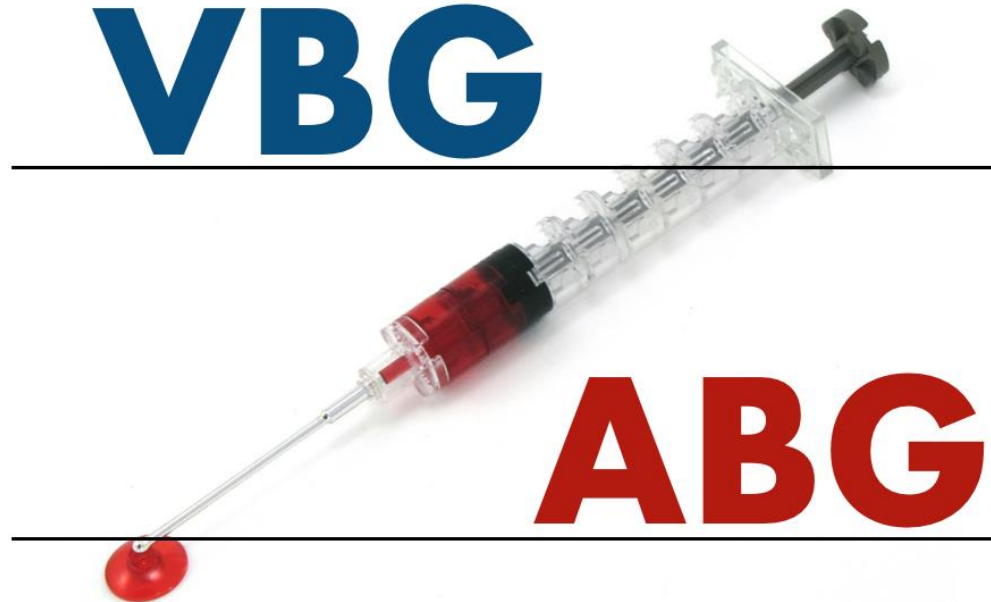
- Sudden dyspnea
- Cyanosis
- Abnormal breath sounds
- Sudden or unexplained tachypnea
- Heavy use of accessory muscles
- Change in ventilator setting
- CPR
- Diffuse infiltrates in c xray



Contraversy

- What do you think about using a “*venous blood gas*”?

VBG



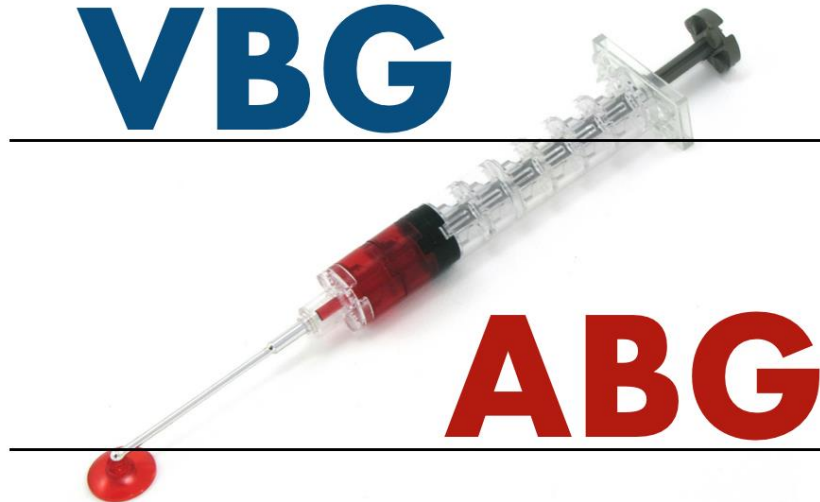
ABG



pH

- Good correlation
- pooled mean difference: +0.035 pH units

VBG



ABG



pCO₂

- Good correlation in normocapnia
- Non-correlative in severe shock
- 100% sensitive in screening for arterial hypercarbia in COPD exacerbation using cut points of PaCO₂ 45 mmHg (6KPA) and lab testing (McCanny et al, 2012)



Bicarbonate

- Good correlation
- Mean difference -1.41 mmol/L (-5.8 to $+5.3$ mmol/L 95%CI)



Lactate

- Poor correlations above 2 mmol/L
- Mean difference 0.08 (-0.27 – 0.42 95%CI)



Base Excess

- Good correlation
- Mean difference 0.089 mmol/L (−0.974 to +0.552 95%CI)



Oxygen

- PO_2 values compare poorly
- arterial PO_2 is typically 36.9 mmHg greater than the venous with significant variability



Clinical Scenarios suited to VBG

- Diabetic Emergencies – i.e. DKA, HHS
- Septic Screening
- Monitor (Hyponatraemia, HB monitoring)
- Possibly Trauma
- ?Others
 - *When might you use this in your practice?*



ABG required if...

- Accurate measurement of PaCO₂ in shock
- Accurate measurement of PaCO₂ if hypercapnic (i.e. PaCO₂ >45 mmHg)
- Accurate Lactate
- PO₂ for planning (Electively)
- PO₂ acutely is saturations problematic
- Equivocal Cases



AUDIT at our centre - *Utilisation of point of care blood gases within 4-hours of ED arrival in trauma patients (ISS \geq 12)*

Recorded blood gas sampling <4h of ED presentation	n/%	30-day Mortality
No ABG or VBG performed	83 (18.2%)	2 (2.4%)
VBG only performed	243 (53.4%)	23 (9.5%)
ABG only performed	63 (13.8%)	9 (14.2%)
ABG and VBG performed	12 (2.6%)	1 (8.3%)
<i>* (ABG and VBG performed)</i>	<i>*54 (11.9%)</i>	<i>4 (7.4%)</i>



OUR TRAUMA DATA (N=176)

Table 2 Comparison of paired ABG and venous blood gas samples (n=176)

Blood gas parameters (variable)	Valid sample values (n)	Pearson correlation between arterial and venous samples	Average (arterial and venous)		Arterial minus venous		P value (t-test of zero bias)	Limits of agreement (LOA) ²¹	
			Median	Range	Mean difference (estimated bias)	SD difference		Lower	Upper
pH	175	0.858	7.36	7.02, 7.59	0.036	0.042	<0.001	-0.048	0.120
Base deficit (mmol/L)	176	0.877	-0.43	-16.5, 5	-1.27	1.54	<0.001	-4.35	1.81
Lactate (mmol/L)	169	0.931	1.90	0.55, 14.7	-0.64	0.61	<0.001	-1.86	0.57
HCO ₃ (mmol/L)	175	0.848	24.5	11, 35	-1.97	1.76	<0.001	-5.49	1.55
Hb (g/dL)	172	0.899	139	88.5, 177	-4.69	7.46	<0.001	-19.61	10.23
Sodium (mmol/L)	169	0.886	139.5	129, 147	-1.08	1.36	<0.001	-3.80	1.64
Potassium (mmol/L)	170	0.683	3.9	2.9, 6.9	-0.19	0.50	<0.001	-1.19	0.81
Ionised calcium (mmol/L)	168	0.833	1.19	0.95, 1.31	0.000	0.030	0.537	-0.060	0.060
Creatinine (μmol/L)	162	0.980	82.5	37, 277	-6.35	5.50	<0.001	-17.35	4.65
Glucose (mmol/L)	172	0.969	7.2	4.3, 31	0.37	0.77	<0.001	-1.17	1.91
PCO ₂ (mm Hg)	175	0.747	44.5	21, 94	-7.59	7.09	<0.001	-21.77	6.59
PO ₂ (mm Hg)	171	0.341	71.9	37.6, 483	116.01	113.79	<0.001	-111.57	343.59
O2 Sats (%)	172	0.133	80.43	53.5, 100	36.74	22.45	<0.001	-8.16	81.64

PCO₂, partial pressure of carbon dioxide; PO₂, partial pressure of oxygen.

10.1136/emered-2020-209751



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Typical Blood Gas – 1 minute

Blood Gas Values

↑ pH	7.451		[7.350 - 7.450]
↑ pCO ₂	54.5	mmHg	[35.0 - 45.0]
↓ pO ₂	58.3	mmHg	[75.0 - 100]

Acid Base Status

cHCO ₃ ⁻ (Pst) _c	35.6	mmol/L
cBase(Ecf) _c	12.7	mmol/L



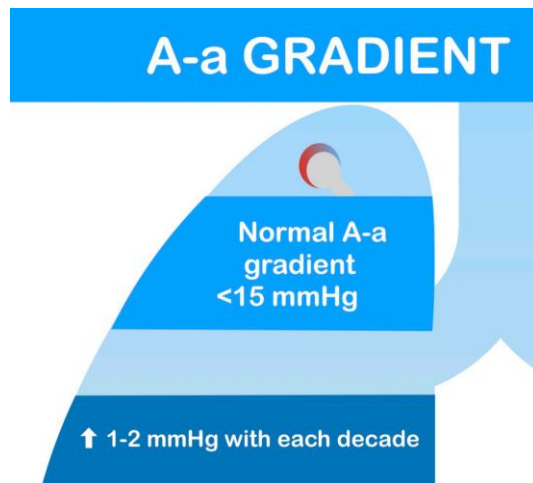
5 Basic Rules – 1 minute ABG

1. How is the patient?

Ask or look...

2. Is the patient hypoxaemic?

*Is the patient relatively hypoxaemic?**



*Quick Rule of thumb x the Oxygen % by 5



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3rd Basic Rules

Is the patient acidotic or alkalotic?
- pH direction generally telling:

If the pH is **normal**, there must be

- 2 or more problems (*mixed picture*)
- no problem
- pregnant patient (*compensated respiratory alkalosis*)



4th and 5th Basic Rules

4. What has happened to the PaCO_2

Is the abnormality wholly or partly due to a defect in the respiratory system?

5. What has happened to the base excess and/or bicarbonate?

Is the abnormality wholly or partly due to a defect in the metabolic system?

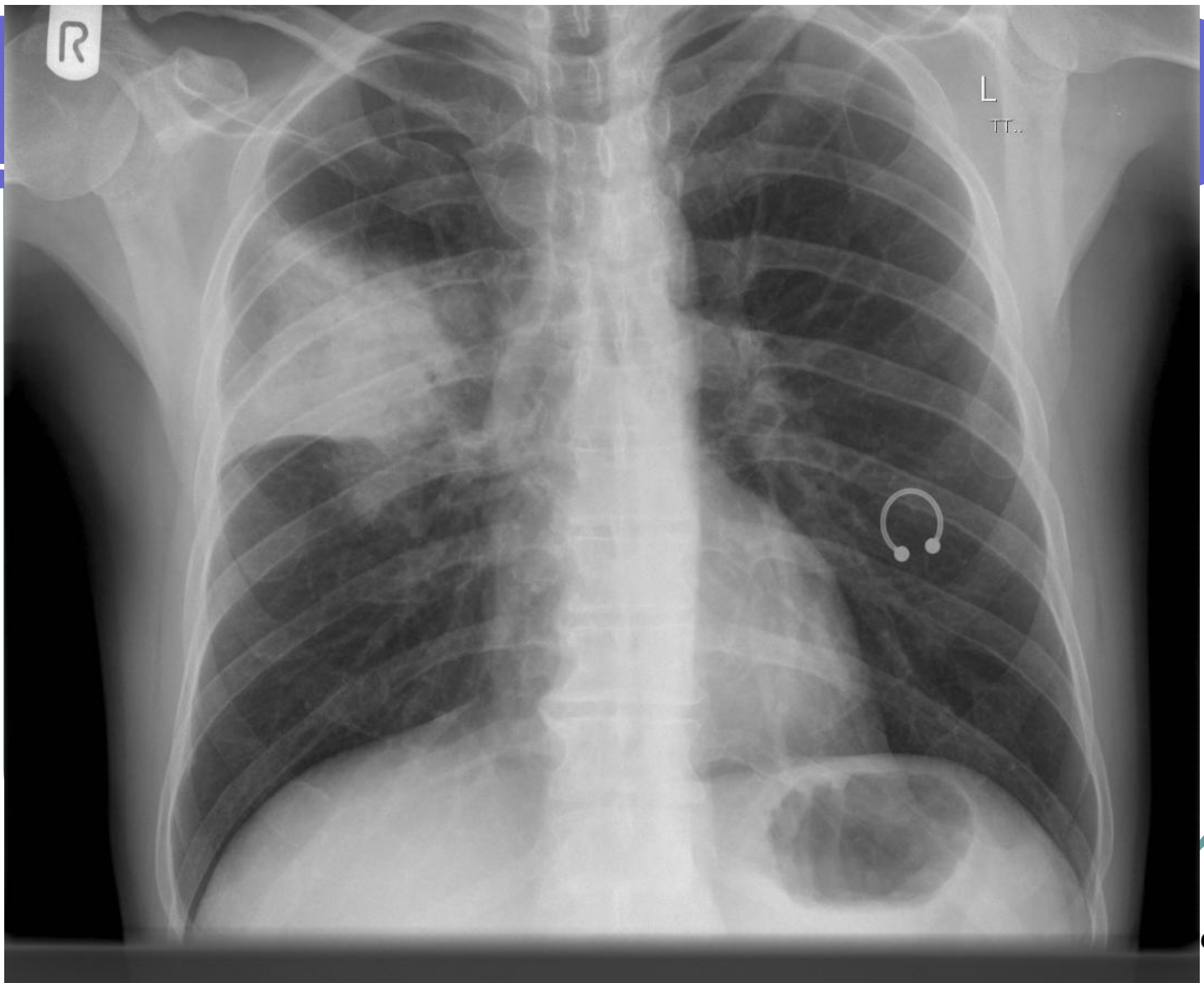


Scenario 1

- 48 year old smoker with fever and cough and shortness of breath.
- He has a poor saturations trace and appears unwell
- Blood cultures have been sent and his venous lactate is 5.1mmol
- You wish to determine his O2 and an accurate lactate

(This is his Chest X-ray):





Scenario 1 - ABG

Blood Gas Values			
↓ pH	7.28		[7.350 - 7.450]
pCO ₂	55.1	mmHg	[32.0 - 46.0]
↓ pO ₂	69.7	mmHg	[83.0 - 108]
cHCO ₃ ⁻ (P) _c	23.3	mmol/L	
cBase(Ecf) _c	- 1.2	mmol/L	
Oximetry Values			
FCOHb	0.81	%	
FMetHb	1.0%	%	
↓ ctHb	86	g/dL	[12.0 - 16.0]
↓ sO ₂		%	[95.0 - 99.0]
Electrolyte Values			
cK ⁺	4.6	mmol/L	[-]
↓ cCl ⁻	---	mmol/L	[98 - 106]
↓ cNa ⁺	137	mmol/L	[136 - 145]



Scenario 1

- How do you determine severity of his pneumonia?
- What treatment(s) would you recommend



Scenario 2

- **65 year woman with sudden onset right sided chest pain and dyspnoea.**
 - What are the clinical risk factors for Pulmonary Embolus (PE)?
 - What is your current diagnostic approach to confirm or exclude PE?
 - What is the role of D dimer, VQ, CTPA, CXR, ECG, blood gases in diagnosing PE?



Scenario 2 – Blood Gas

- **This is her arterial blood gas (ABG) result on no oxygen (21%):**
 - pH 7.51, PO₂ 49, PCO₂ 27, HCO₃ 23, Base Excess -1.9
 - How do you interpret this ABG result?



Scenario 3

- **19 year old woman with history of asthma presents with dyspnoea?**
- **Resp Rate 45, HR 140, BP 130/70**
- How would you assess the severity of her current asthmatic attack?
- What is the role of a blood gas in Asthma?

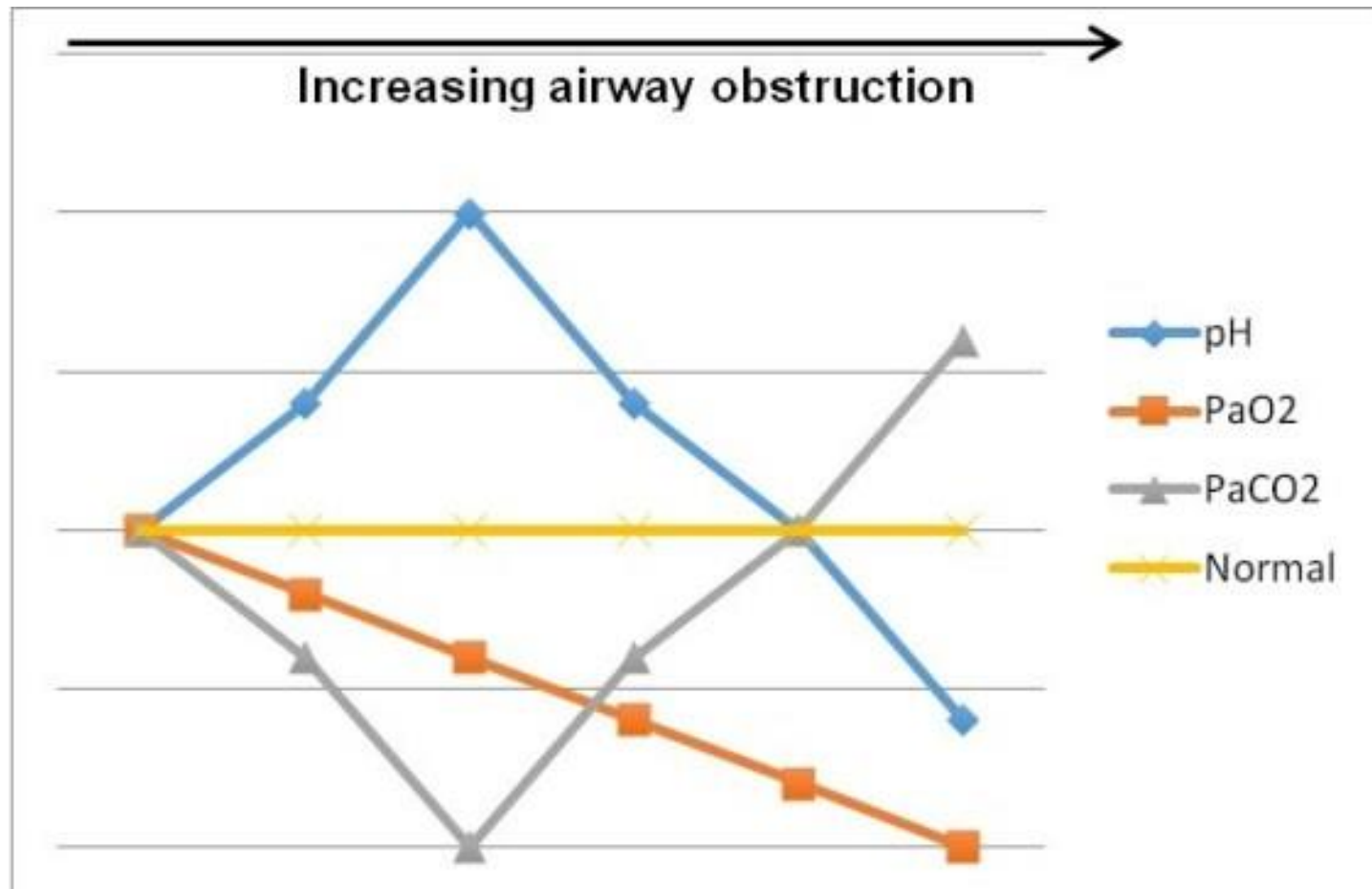


Here is her blood gas (venous)

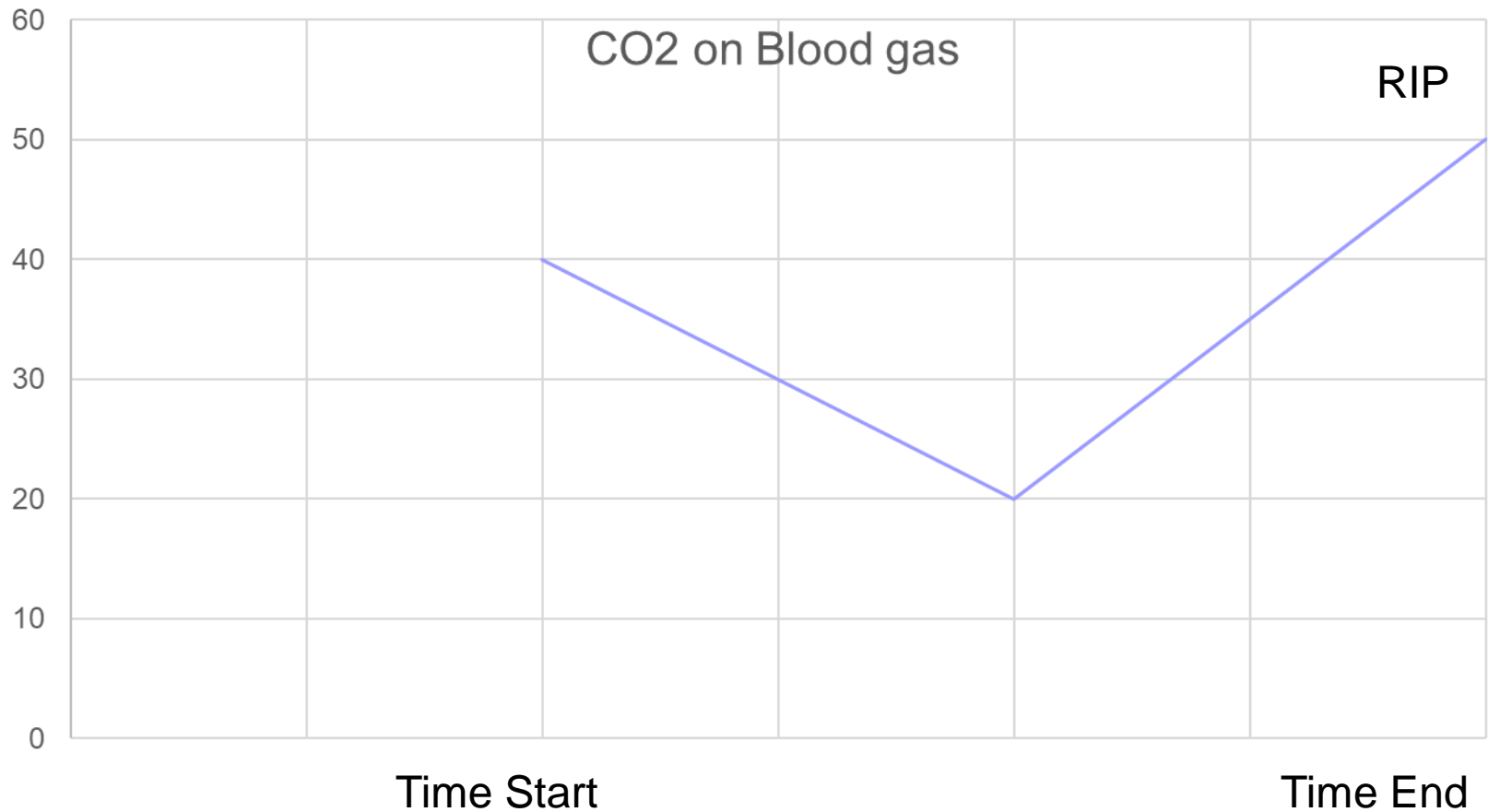
- pH 7.40
- PaCO₂ 42
- PaO₂ 150
- Bicarbonate 23
- B/E -1



CO₂ in Asthma



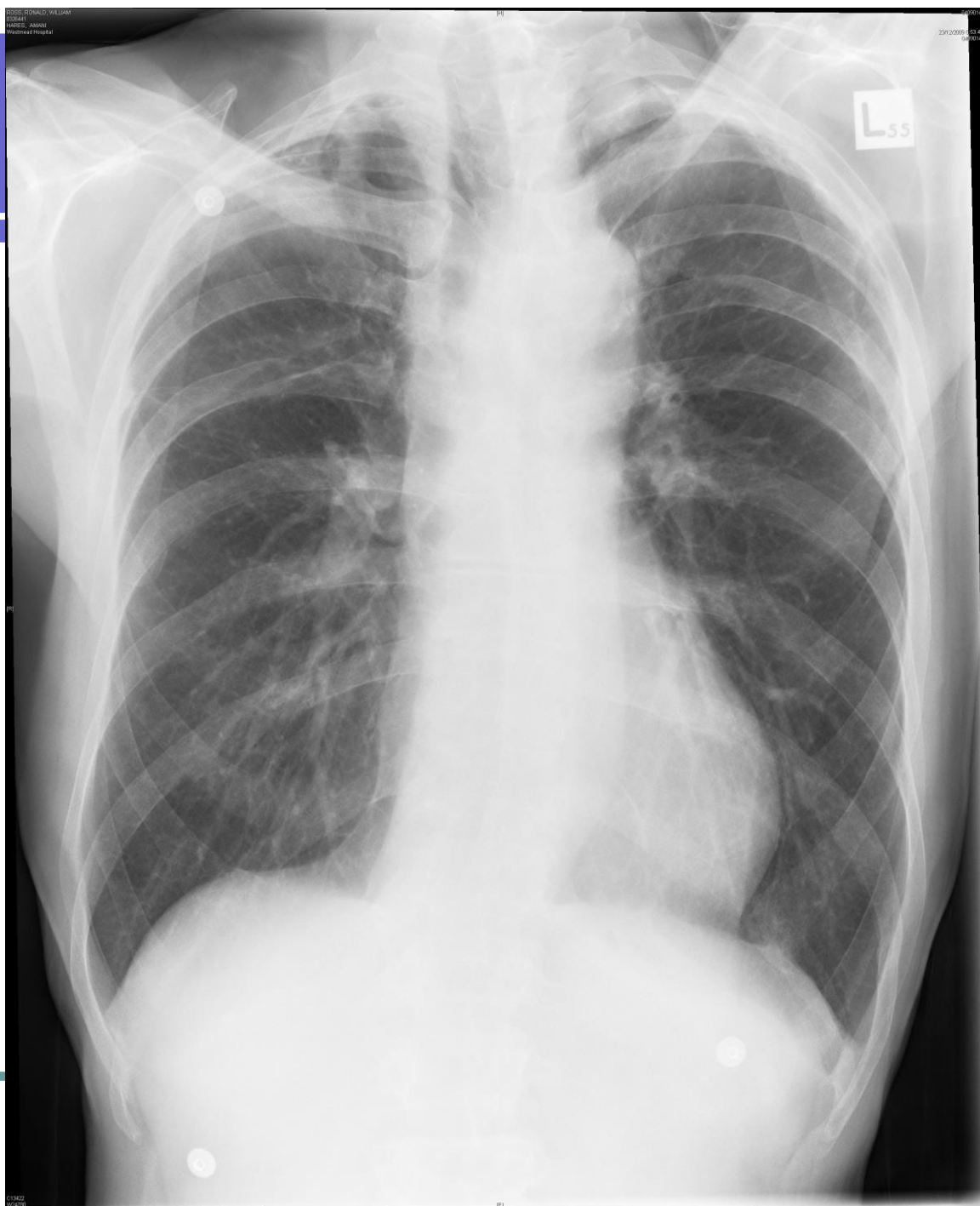
Time v Co2



Scenario 4

- **74 year old man, life-long smoker, with severe respiratory distress and a saturation of 82%.**
- This is his Chest X-ray.





Scenario 4

- What factors could have exacerbated his chronic airway limitation (CAL)?
- How do you manage him initially?



ABG

Blood Gas Values

↓ pH	7.282		[7.350 - 7.450]
↑ $p\text{CO}_2$	55.1	mmHg	[35.0 - 45.0]
↓ $p\text{O}_2$	69.7	mmHg	[75.0 - 100]

Acid Base Status

$\text{cHCO}_3^- (\text{P,st})_c$	23.3	mmol/L	
cBase(B)_c	-1.2	mmol/L	[-3.0 - 3.0]

Electrolyte Values

cK^+	4.6	mmol/L	[3.4 - 5.5]
cNa^+	137	mmol/L	[136 - 146]
↓ cCa^{2+}	1.08	mmol/L	[1.15 - 1.30]
$\text{cCa}^{2+} (7.4)_c$	1.01	mmol/L	
cCl^-	100	mmol/L	[94 - 107]

Metabolite Values

↑ cGlu	6.0	mmol/L	[3.9 - 5.8]
↑ cLac	2.7	mmol/L	[0.5 - 2.0]



Should we remove the O2



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covid-19

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Practice » Lesson of the Week

Rebound hypoxaemia after administration of oxygen in an acute exacerbation of chronic obstructive pulmonary disease

BMJ 2011 ; 342 doi: <https://doi.org/10.1136/bmj.d1557> (Published 31 March 2011)

Cite this as: BMJ 2011;342:d1557

Article

Related content

Metrics

Responses

Binita Kane, SpR in Respiratory Medicine¹, Peter M Turkington, consultant physician in respiratory medicine¹,



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Scenario 5

Rebound Hypoxia - described in BMJ

- Consider well lady with CAL
pH 7.40 c
CO₂ 34
O₂ 60
Bicarbonate 24
- Develops exacerbation CAL and given 4litres/min NP for days
pH 7.22
CO₂ 90
O₂ 150mmhg
Bicarbonate 36
- **Oxygen removed**

Oxygen therapy in COPD

- 20 mins after O₂ removed

pH 7.28

CO₂ 82

O₂ 32 YES 32 !

Bicarb 37

WHY? CO₂ stores are extensive

Put this in your alveolar gas equation

$PAO_2 = FiO_2 (760 - 47) - pCO_2 / R0.8$

147- 102 only 45 at best without considering abnormal lungs

24% would give at best 60 but she has a big a/A gradient

Needed 28% at least to prevent rebound hypoxia

What happened?

- Rebound – Why?
 - Body's CO_2 stores are very large
 - Stop the oxygen, pO_2 drops quickly
- pCO_2 does not, especially if very high – in this case the alveolar oxygen pressure can drop below where it was before O_2 commenced

The Alveolar Gas Equation:

$$P_A O_2 = P_i O_2 - \frac{P_a CO_2}{0.8}$$

$$P_i O_2 = F_i O_2 (P_b - P_{H_2O})$$



Oxygen administration

Acute deterioration: hypoxia kills *not* hypercarbia

Chronic (i.e. COPD)

SaO₂ 88-92%

Start FiO₂ low; ☒ O₂ to
maintain SaO₂

Monitor CO₂

NEVER *abruptly* remove O₂
Rebound hypoxaemia. BMJ ref

Acute (i.e. Asthma)

SaO₂ 94-98%

Hudson 6-8ltr

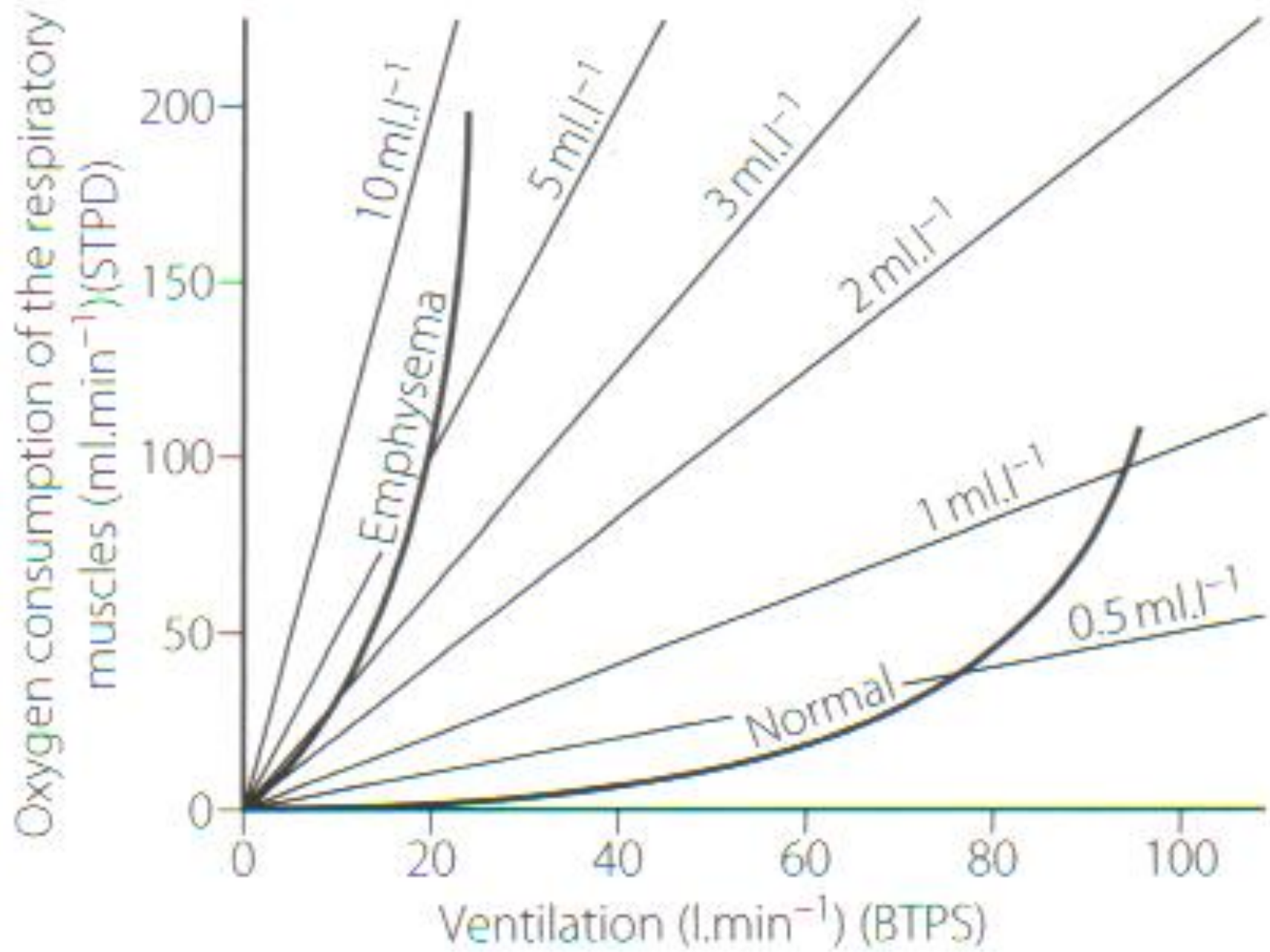
Beware ☒ CO₂
= exhaustion
HELP

Do not *abruptly*
remove O₂



Severe COPD

- Tachypnoea
- Marked dyspnoea
- Pursed lip breathing
- Use of accessory muscles at rest
- Acute confusion
- New onset cyanosis
- New onset peripheral oedema
- Marked reduction in ADL's



COPD

- Up to 25% of patients with a severe unexplained exacerbation of COPD may have a co-existing PE.

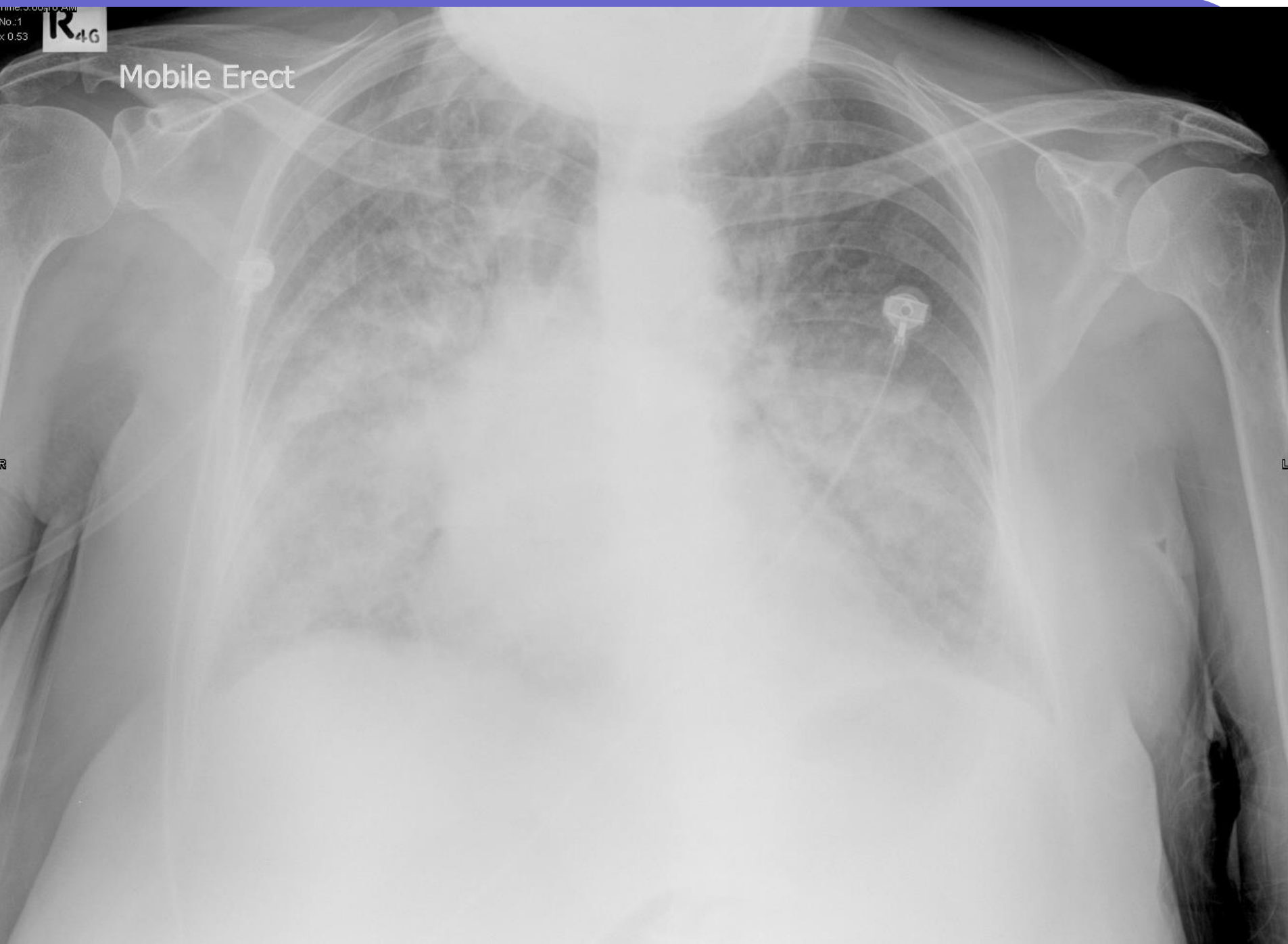
• *Tillie-Leblond, Marquette et al, Annals of Internal Medicine 2006*

Scenario 6

- **67 year old woman in pre planning for dialysis and with severe heart failure is brought in by ambulance at 5am with sudden onset respiratory distress.**
- **This is her Chest X-ray**



Mobile Erect



Scenario 6

- What are the conflicting issues in managing this patient?
- Please interpret their ABG



What is your interpretation

			Reference range
pH	7.31		(7.35-7.45)
PCO ₂	30	mmHg	(35-45)
PO ₂	104	mmHg	(75-100)
HCO ₃	18.5	mmol/L	(22-33)
BE	-4.8		(-3.0-+3.0)
Saturation	99%		(95-98%)
FIO ₂	0.21		
Na ⁺	141	mmol/L	(135-145)
K ⁺	8.4	mmol/L	(3.2-4.5)
Ca ²⁺	1.21	mmol/L	(1.15-1.35)
Cl ⁻	113	mmol/L	(100-110)





Advanced Rules

Shortcut:

$$\text{PAO}_2 \approx \text{FiO}_2 \times 500$$

$$\text{PAO}_2 = \text{FiO}_2(\text{P}_B - \text{P}_{\text{H}_2\text{O}}) - (\text{PaCO}_2/\text{RQ})$$

P_B is barometric pressure

$\text{P}_{\text{H}_2\text{O}}$ is pressure due to water vapour

RQ is respiratory quotient

Breathing room air at sea level:

$$\begin{aligned}\text{PAO}_2 &= 0.21 \times (760 - 47) - (40/0.8) \\ &= 150 - 100 \\ &= 100 \text{ mmHg}\end{aligned}$$

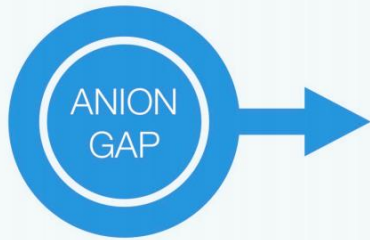


Acidosis Rules

$$\text{ANION GAP} = [\text{Na}] - [\text{HCO}_3] - [\text{Cl}]$$



Acidosis Rules



Rule 1

If you see a metabolic acidosis, you **must** calculate the **anion gap**

$$\text{ANION GAP} = [\text{Na}] - [\text{HCO}_3] - [\text{Cl}]$$

Normal 12 (range 6-15)
Albumin correction = $\text{AG} + \frac{1}{4}(44 - \text{albumin})$



Rule 2

If the anion gap is elevated, you should calculate the **delta ratio**

$$\text{DELTA RATIO} = \frac{\uparrow \text{ in AG}}{\downarrow \text{ in } [\text{HCO}_3]}$$

$$\text{DELTA RATIO} = \frac{\text{AG} - 12}{24 - [\text{HCO}_3]}$$

<0.8 = combined HAGMA & NAGMA
1-2 = uncomplicated HAGMA
>2 = pre-existing metabolic alkalosis



Rule 3

If you see a measured osmolality, you **must** calculate the **osmolar gap**

$$\text{OSMOLAR GAP} = \text{osmolality} - \text{osmolarity}$$

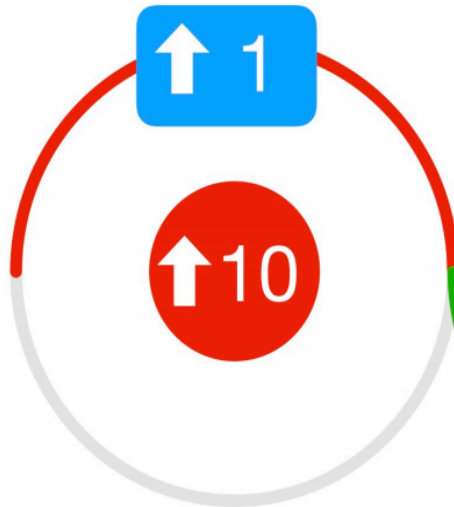
osmolality is *measured*
osmolarity is *calculated*
calc osmolarity =
 $2[\text{Na}] + \text{urea} + \text{glucose}$



COMPENSATION RULES



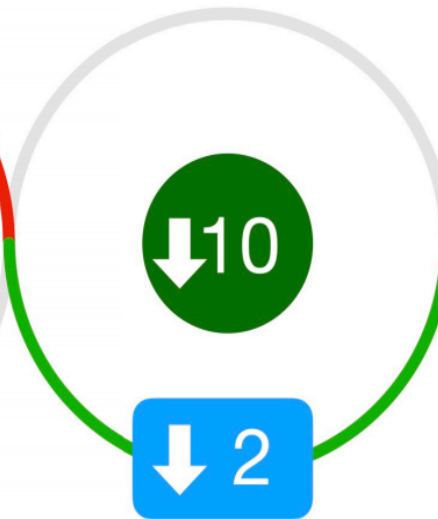
Acute
Respiratory
Acidosis



HCO_3^- \uparrow 1 mmol for every
10mmHg \uparrow in $\text{CO}_2 > 40$

Acute ACidosis
1 for 10

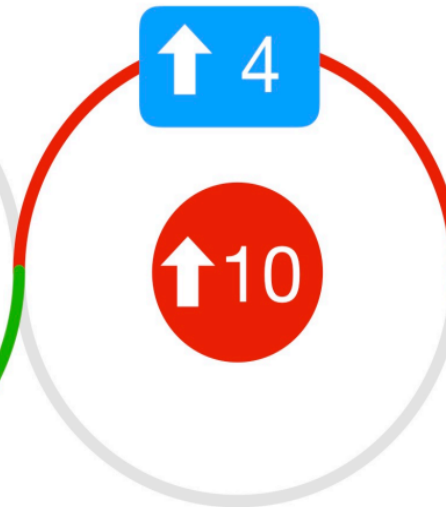
Acute
Respiratory
Alkalosis



HCO_3^- \downarrow 2 mmol for every
10mmHg \downarrow in $\text{CO}_2 < 40$

Acute ALkalosis
2 for 10

Chronic
Respiratory
Acidosis



HCO_3^- \uparrow 4 mmol for every
10mmHg \uparrow in $\text{CO}_2 > 40$

Chronic ACidosis
4 for 10

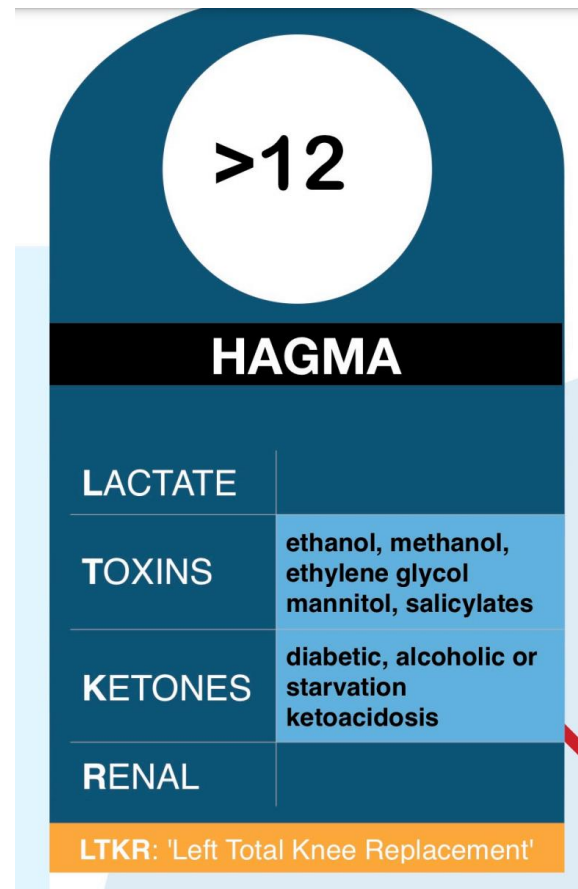
Chronic
Respiratory
Alkalosis



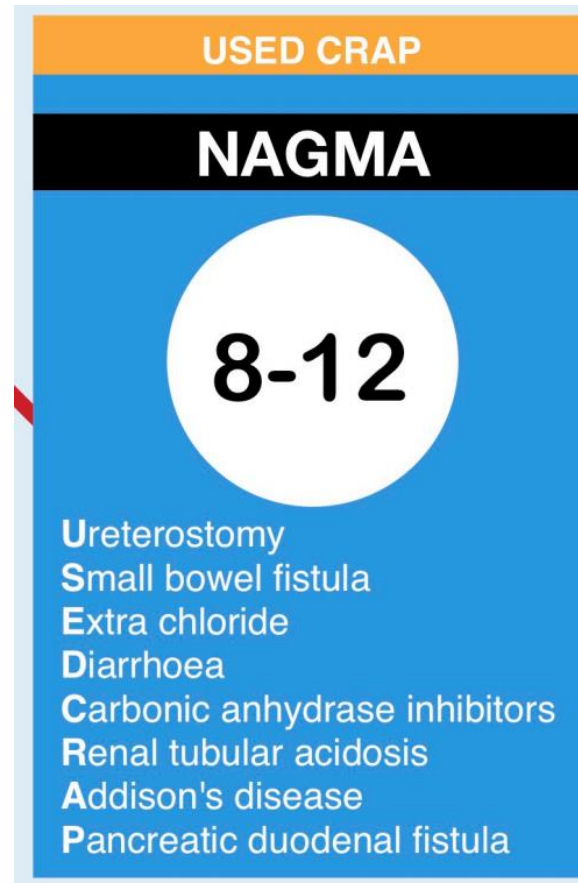
HCO_3^- \downarrow 5 mmol for every
10mmHg \downarrow in $\text{CO}_2 < 40$

Chronic ALkalosis
5 for 10

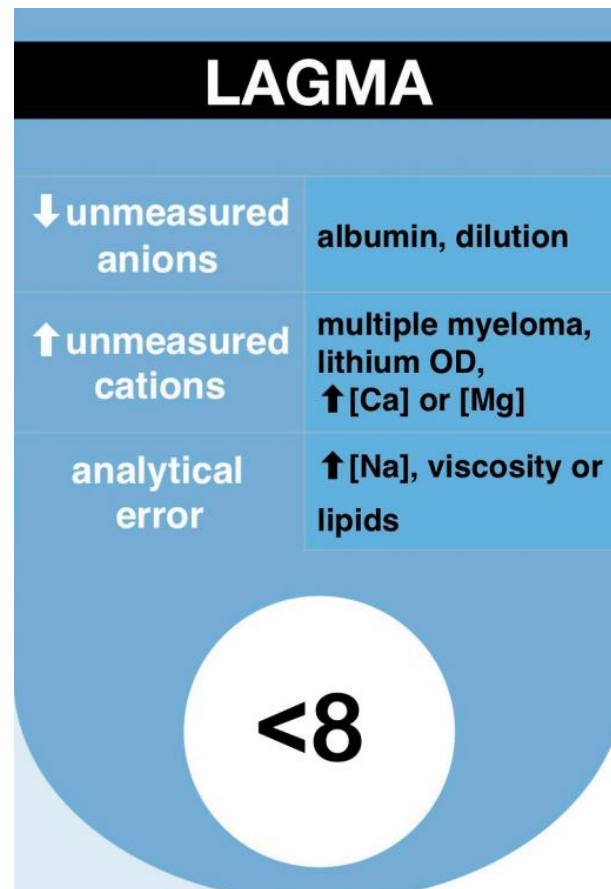
Acidosis Rules



Acidosis Rules



Acidosis Rules



Osmolar Gap

OSMOLAR GAP
=
osmolality
- osmolarity

>10

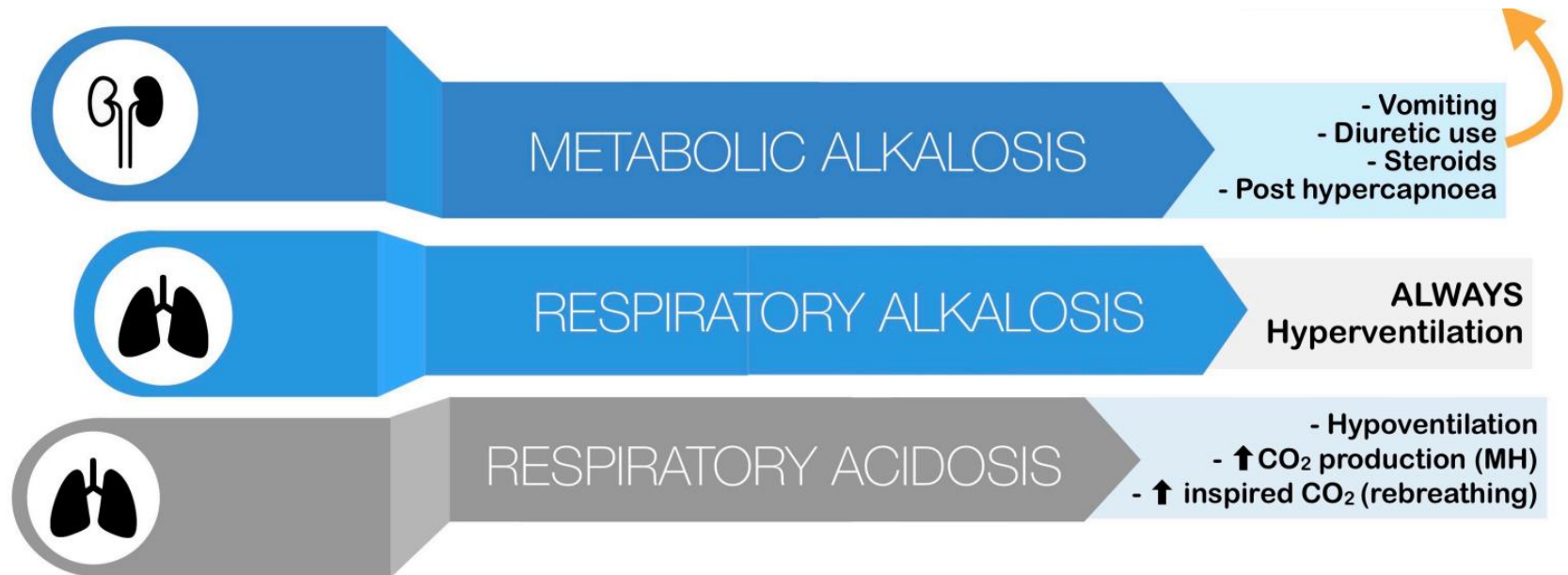
MIME ELK

**R
A
I
S
E
D**

Methanol/mannitol
Isopropyl alcohol
Methylene glycol
Ethanol
Ethylene glycol
Lactate
Ketones



Other Problems



Blood Gas Challenge 1

Serum biochemistry and arterial blood gas

FiO ₂	0.50	
pH	7.05	
pCO ₂	66	mmHg
pO ₂	247	mmHg
Bicarbonate	18	mmol/L
Base excess	-14	
O ₂ saturation	99	%
Na ⁺	131	mmol/L
K ⁺	5.0	mmol/L
Cl ⁻	92	mmol/L
Urea	15	mmol/L
Creatinine	227	micromol/L
Glucose	50.9	mmol/L

Reference Range

7.35-7.45
35-45
80-95
22-28
-3 - +3

Rule 2

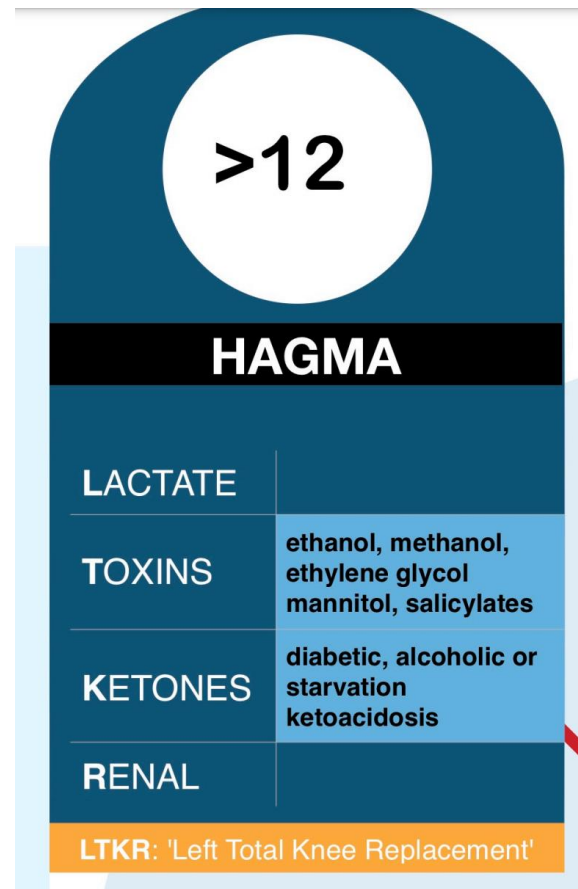
If the anion gap is elevated,
you should calculate the
delta ratio

$$\text{DELTA RATIO} = \frac{\uparrow \text{ in AG}}{\downarrow \text{ in } [\text{HCO}_3^-]}$$

$$\text{DELTA RATIO} = \frac{\text{AG} - 12}{24 - [\text{HCO}_3^-]}$$

<0.8 = combined HAGMA & NAGMA
1-2 = uncomplicated HAGMA
>2 = pre-existing metabolic alkalosis

Acidosis Rules



Blood Gas Challenge 2

FiO ₂	0.21		
pH	7.24		7.35-7.45
pCO ₂	92	mmHg	35-45
pO ₂	45	mmHg	80-95
Bicarbonate	49	mmol/L	22-28
Base excess	10		-3 - +3
O ₂ saturation	78	%	> 95
Lactate	1.2	mmol/L	< 1.3
Na ⁺	142	mmol/L	134-146
K ⁺	3.8	mmol/L	3.4-5
Cl ⁻	86	mmol/L	98-106
Glucose	11.4	mmol/L	3.5-5.5
Haemoglobin	184	g/L	135-180
Carboxy Hb	7	%	< 6%



Blood Gas Challenge 3

Bedside venous blood gas results included:

			Reference Range
FIO ₂	0.21		
pH	7.25		7.35-7.45
pCO ₂	35	mmHg	35-45
pO ₂	234	mmHg	80-95
HCO ₃	15.0	mmol/L	22-28
Base Excess	-11		-3 to +3
Na ⁺	141	mmol/L	134-146
K ⁺	1.1	mmol/L	3.4-5.0
Cl ⁻	116	mmol/L	98-106
Glucose	5.4	mmol/L	3.5-4.5



Answers

What is her main acid/base disorder?

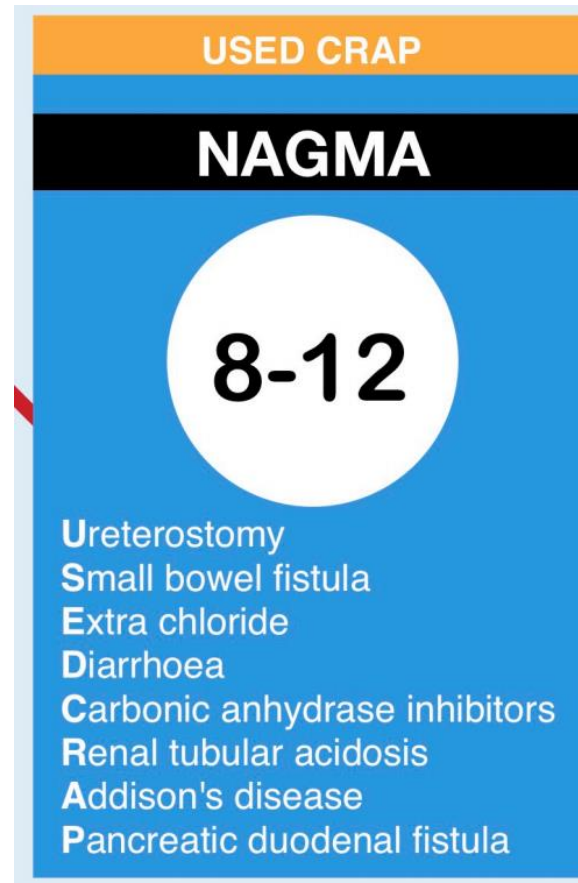
Metabolic acidosis: low pH +
normal CO₂ + normal HCO₃ +
strongly negative base excess.

What is her anion Gap?

$$\begin{aligned} &\text{Na} - [\text{HCO}_3 - \text{Cl}] \\ &= 141 - 15 - 116 = 10 \end{aligned}$$



Acidosis Rules



Blood Gas Challenge 4

A 26 year old mechanic is brought in by ambulance after an overdose of an unknown substance. On arrival his vital signs are:

HR 106
BP 80/50
GCS 12
sats 100% on RA
RR 34



ABG

FiO2	0.21	
pH	7.15	
pO ₂	115	mmHg
pCO ₂	20	mmHg
HCO ₃ ⁻	8	mmol/L
BE	-16	
Na	135	mmol/L
K	6.5	mmol/L
Cl	95	mmol/L
glucose	6.2	mmol/L
urea	8.1	mmol/L
lactate	5.2	mmol/L
measured osmolality	320	mosm/Kg

Rule 3

If you see a measured osmolality, you **must** calculate the **osmolar gap**

OSMOLAR GAP = osmolality - osmolarity

osmolality is *measured*
osmolarity is *calculated*
calc osmolarity =
 $2[\text{Na}] + \text{urea} + \text{glucose}$



Osmolar Gap

OSMOLAR GAP
=
osmolality
- osmolarity

>10

MIME ELK

**R
A
I
S
E
D**

Methanol/mannitol
Isopropyl alcohol
Methylene glycol
Ethanol
Ethylene glycol
Lactate
Ketones

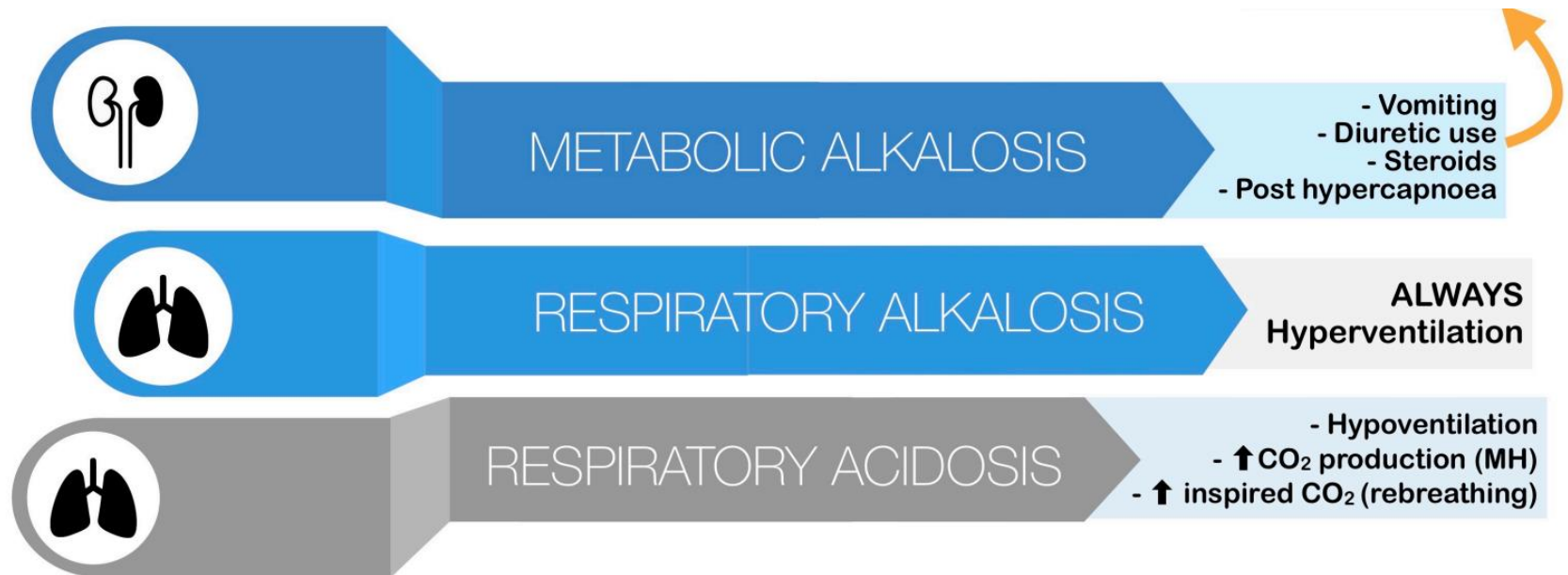


Blood Gas Challenge 5

Blood Gas Values				
↑ pH	7.451		[7.350 - 7.450]	
↑ pCO ₂	54.5	mmHg	[35.0 - 45.0]	
↓ pO ₂	58.3	mmHg	[75.0 - 100]	
Acid Base Status				
cHCO ₃ ⁻ (P,st) _c	35.6	mmol/L		
cBase(Ecf) _c	12.7	mmol/L		
Oximetry Values				
↓ sO ₂	88.5	%	[95.0 - 100.0]	
↓ ctHb	99	g/L	[130 - 180]	
Electrolyte Values				
↓ cK ⁺	2.9	mmol/L	[3.5 - 5.0]	
cNa ⁺	144	mmol/L	[136 - 146]	
↓ cCa ²⁺	1.14	mmol/L	[1.15 - 1.30]	
cCa ²⁺ (7.4) _c	1.17	mmol/L	[- -]	
Metabolite Values				
cGlu	5.4	mmol/L	-	
cLac	1.5	mmol/L	0.5 - 2.0	
Temperature Corrected Values				
pH(T)	7.451			
pCO ₂ (T)	54.5	mmHg		



Other Problems



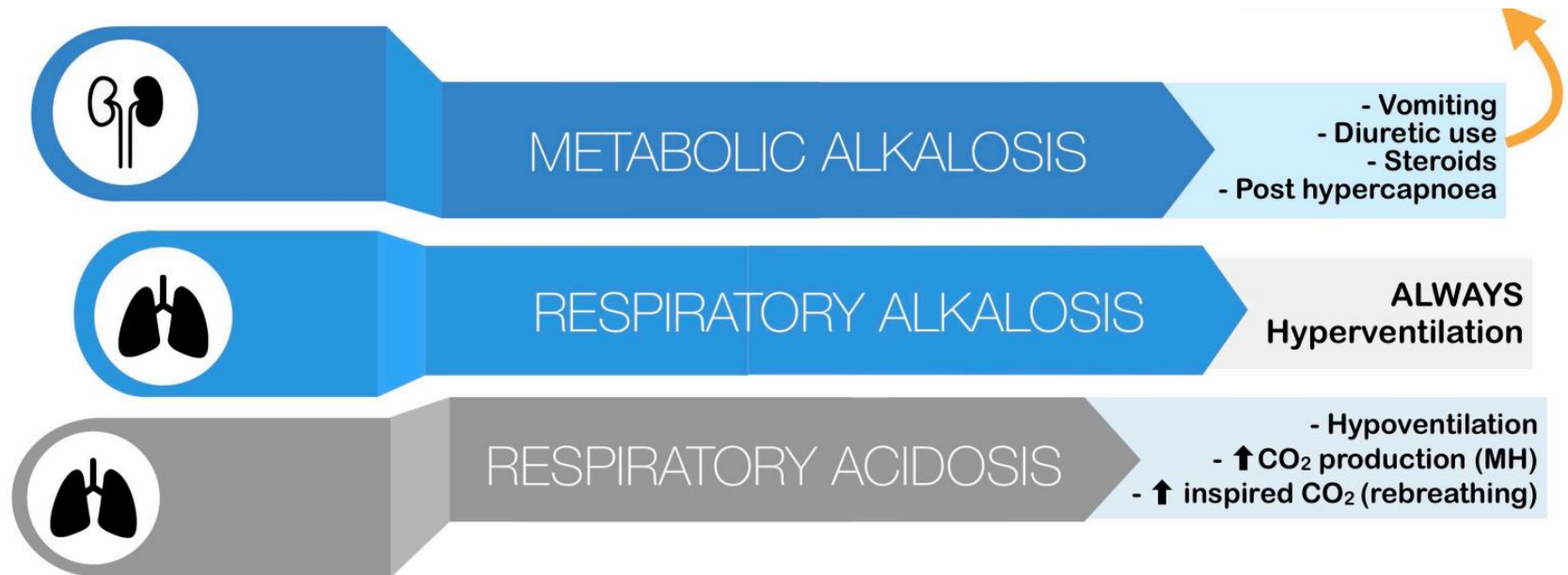
Blood Gas Challenge 6

Reference Range

FIO ₂	0.5		
pH	7.62		(7.35-7.45)
pCO ₂	28.5	mmHg	(35-45)
pO ₂	234	mmHg	(80-95)
Bicarbonate	30.0	mmol/L	(22-28)
Base excess	8.3		(-3 - +3)
O ₂ saturation	99.8	%	(> 95)
Lactate	1.1	mmol/L	(< 1.3)
Na ⁺	131	mmol/L	(134-146)
K ⁺	2.0	mmol/L	(3.4-5)
Cl ⁻	90	mmol/L	(98-106)
Glucose	12.7	mmol/L	(3.5-5.5)



Other Problems



The End.

